<u>GENERAL TECHNOLOGIES, SPC</u>

- High-Quality Services & Products

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D850 - Na CHELATING WEAK ACID CATION EXCHANGE RESIN

(Designed for selective heavy metal removal in wastewater treatment applications)

Product Description

D850 (Na form) resin is a premium grade, macroporous weak acid cation exchange resin with iminodiacetate functional groups. It is designed to chelate/remove many heavy metal cations, even in waters containing high calcium concentrations. It is suitable for use in polishing metal cations from wastewater and process water streams.

The relative affinity of D850 for various divalent cations of the following metals in near neutral solutions are as follows:

H>Cu>V>U02>Pb>Ni>Zn>Co>C d>Fe>Be>Mn>Mg>Ca>Sr>Ba

High concentrations of sulfates and chlorides or the presence of chelating or complexing agents can alter this sequence and will affect the operating capacity.

Typical Physical, Chemical & Operating Characteristics

Polymer Structure Polystyrene cross linked with

Divinylbenzene

Physical Form and Appearance amber spherical beads

Whole Bead Count 93% Min.

Functional Groups R-CH₂-N(CH₂COOH)₂

(Iminodiacetic acid)

Ionic Form (as shipped) Na-

Shipping Weight, approx. 720 g/l (45 lb./ft.³)

Mesh Size (U.S. Std) 16-50

Moisture retention, Na+ form 46–52%

Total Capacity in sodium form >1.0 meg/ml

pH Range, operating 1-6

CHEMICAL AND THERMAL STABILITY

D850 resin is insoluble in dilute or moderately concentrated acids, alkalies, and in all common solvents. However, exposure to significant amounts of free chlorine, "hypochlorite" ions, or other strong oxidizing agents over long periods of time will eventually break down the crosslinking. This will tend to increase the moisture retention of the resin, decreasing it s mechanical strength, as well as generating small amounts of extractable breakdown products. The product is thermally stable to higher than $100 \, {}_{\circ}\text{C}$ ($212 \, {}_{\circ}\text{F}$) in the sodium form, and $60 \, {}_{\circ}\text{C}$ ($140 \, {}_{\circ}\text{F}$) in the hydrogen form.